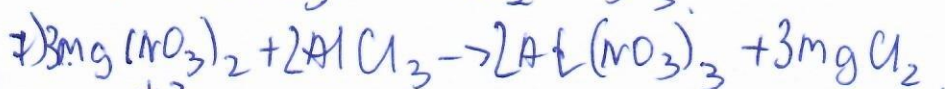
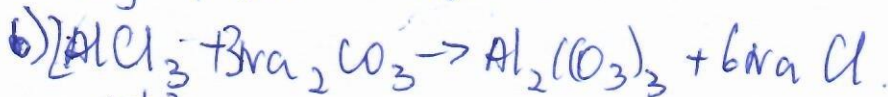
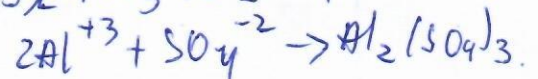
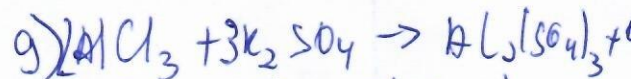
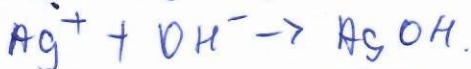
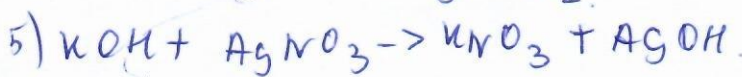
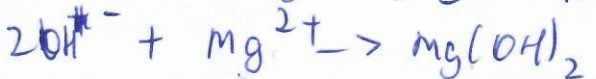
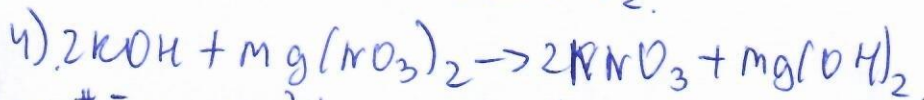
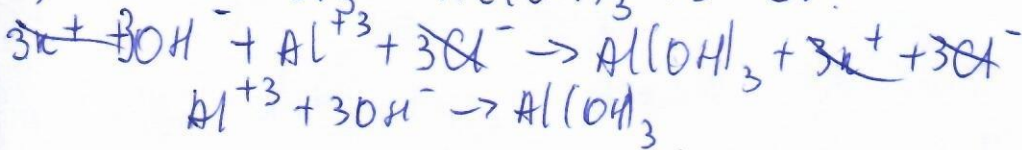
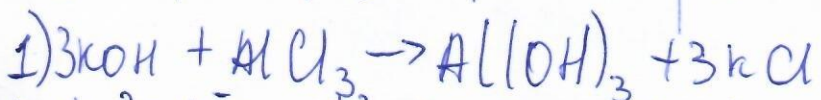
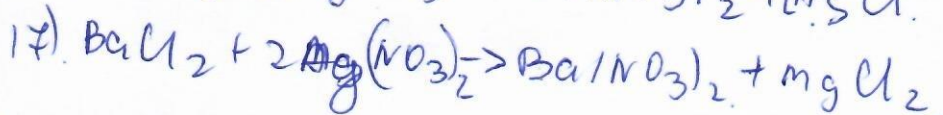
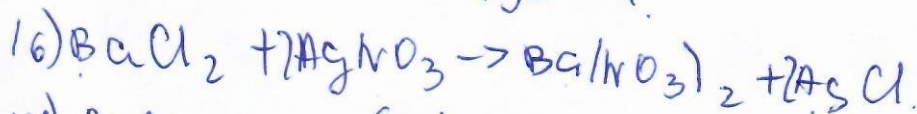
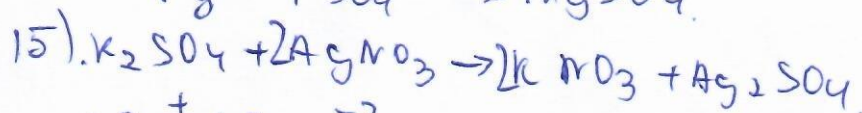
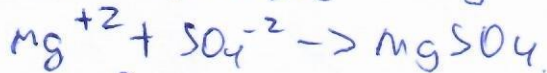
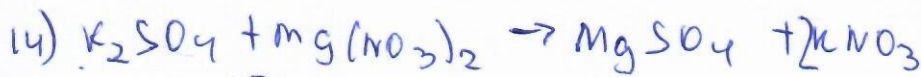
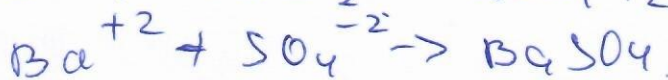
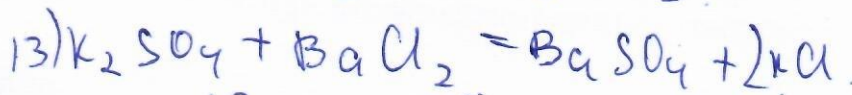
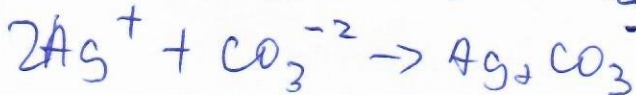
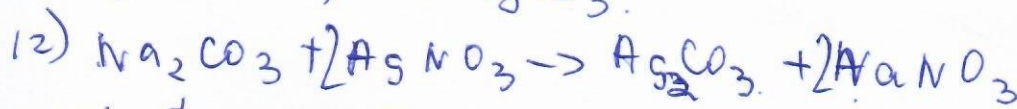
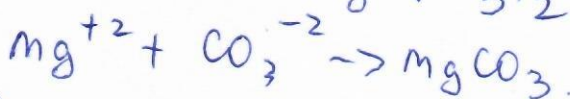
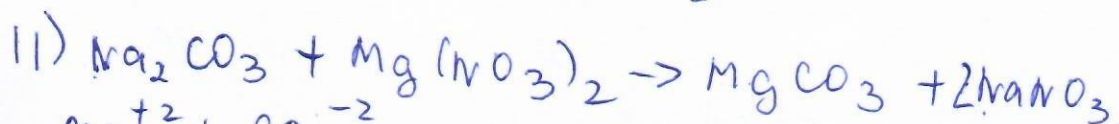
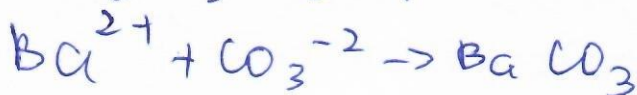
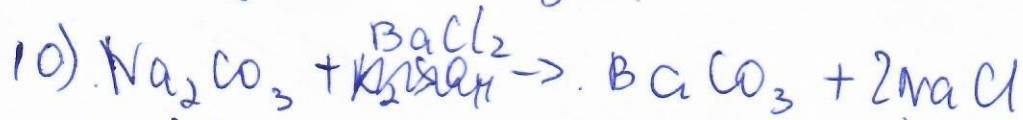
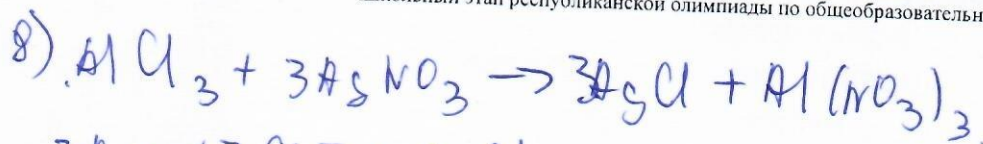


|                                   |   |   |   |   |                             |   |                                 |
|-----------------------------------|---|---|---|---|-----------------------------|---|---------------------------------|
| <del>KOH</del>                    | KOH                                     | AlCl <sub>3</sub>                               | Na <sub>2</sub> CO <sub>3</sub>                 | K <sub>2</sub> SO <sub>4</sub>                  | BaCl <sub>2</sub>           | Mg(NO <sub>3</sub> ) <sub>2</sub>         | AgNO <sub>3</sub>               |
| KOH                               | <del>KOH</del>                          | Al(OH) <sub>3</sub>                             | <del>Na<sub>2</sub>CO<sub>3</sub></del>         | <del>K<sub>2</sub>SO<sub>4</sub></del>          | Ba(OH) <sub>2</sub>         | Mg(OH) <sub>2</sub>                       | AgOH                            |
| AlCl <sub>3</sub>                 | Al(OH) <sub>3</sub>                     | <del>AlCl<sub>3</sub></del>                     | Al <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> | Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> | <del>BaCl<sub>2</sub></del> | MgCl <sub>2</sub>                         | AgCl                            |
| Na <sub>2</sub> CO <sub>3</sub>   | <del>Na<sub>2</sub>CO<sub>3</sub></del> | Al <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> | <del>Na<sub>2</sub>CO<sub>3</sub></del>         | <del>K<sub>2</sub>SO<sub>4</sub></del>          | BaCO <sub>3</sub>           | MgCO <sub>3</sub>                         | Ag <sub>2</sub> CO <sub>3</sub> |
| K <sub>2</sub> SO <sub>4</sub>    | <del>K<sub>2</sub>SO<sub>4</sub></del>  | Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> | <del>Na<sub>2</sub>CO<sub>3</sub></del>         | <del>K<sub>2</sub>SO<sub>4</sub></del>          | BaSO <sub>4</sub>           | MgSO <sub>4</sub>                         | Ag <sub>2</sub> SO <sub>4</sub> |
| BaCl <sub>2</sub>                 | Ba(OH) <sub>2</sub>                     | <del>AlCl<sub>3</sub></del>                     | BaCO <sub>3</sub>                               | BaSO <sub>4</sub>                               | <del>BaCl<sub>2</sub></del> | MgCl <sub>2</sub>                         | AgCl                            |
| Mg(NO <sub>3</sub> ) <sub>2</sub> | Mg(OH) <sub>2</sub>                     | MgCl <sub>2</sub>                               | MgCO <sub>3</sub>                               | MgSO <sub>4</sub>                               | MgCl <sub>2</sub>           | <del>Mg(NO<sub>3</sub>)<sub>2</sub></del> | <del>AgNO<sub>3</sub></del>     |
| AgNO <sub>3</sub>                 | AgOH                                    | AgCl  | Ag <sub>2</sub> CO <sub>3</sub>                 | Ag <sub>2</sub> SO <sub>4</sub>                 | AgCl                        | <del>Mg(NO<sub>3</sub>)<sub>2</sub></del> | <del>AgNO<sub>3</sub></del>     |



X-11-2

Жалпы білім беретін пәндер бойынша республикалық олимпиаданың мектепшілік кезеңі  
Школьный этап республиканской олимпиады по общеобразовательным предметам



Бұл реакциялерде барлық реагенттерді бір-бірімен арақатынастағында  
реакция түзіледі. 7 түрлі координациялық құрылымдар бар.  
Түзілім түрлері:  $\text{Al}(\text{OH})_3$ ,  $\text{Ba}(\text{OH})_2$ ,  $\text{Mg}(\text{OH})_2$ ,  $\text{AgOH}$  т.б.

X-11-2

